

# Understanding the risks of Hydrogen

The insurance perspective

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September 2022  
Risk Engineering Services

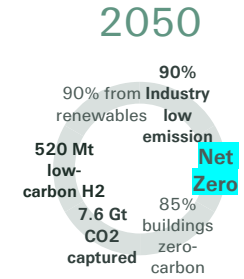
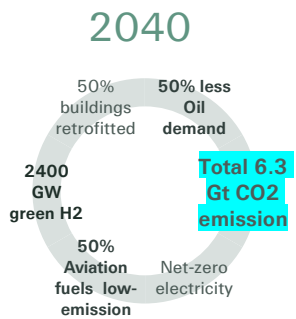
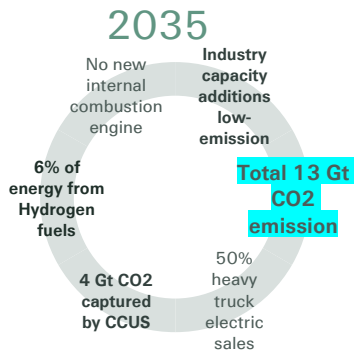


# Understanding the risks of Hydrogen

## A small molecule with a big potential

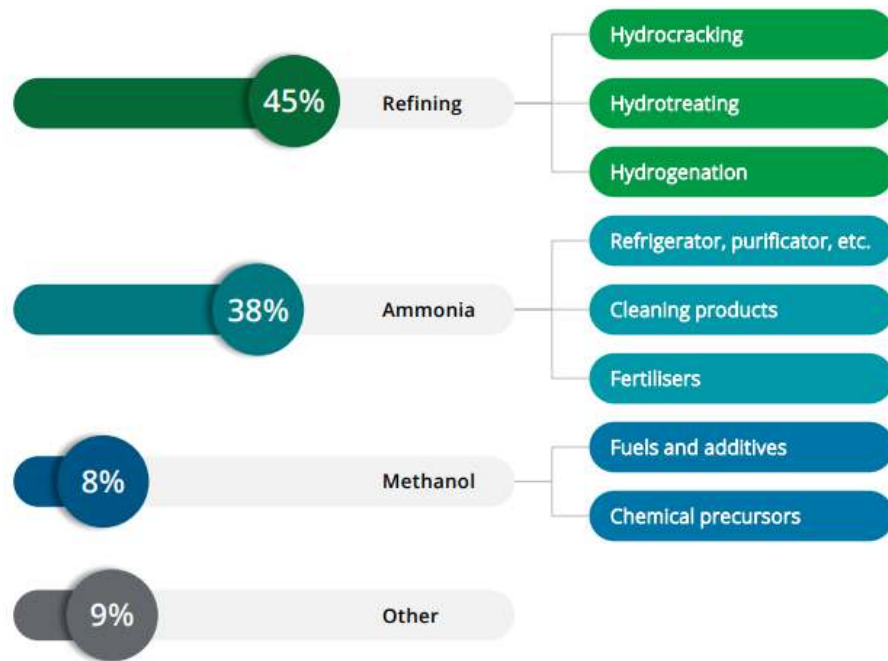


**Hydrogen** means  
 "Creator (-gen) of water (hydro-):  
 its combustion releases only water



# Understanding the risks of Hydrogen Hydrogen demand

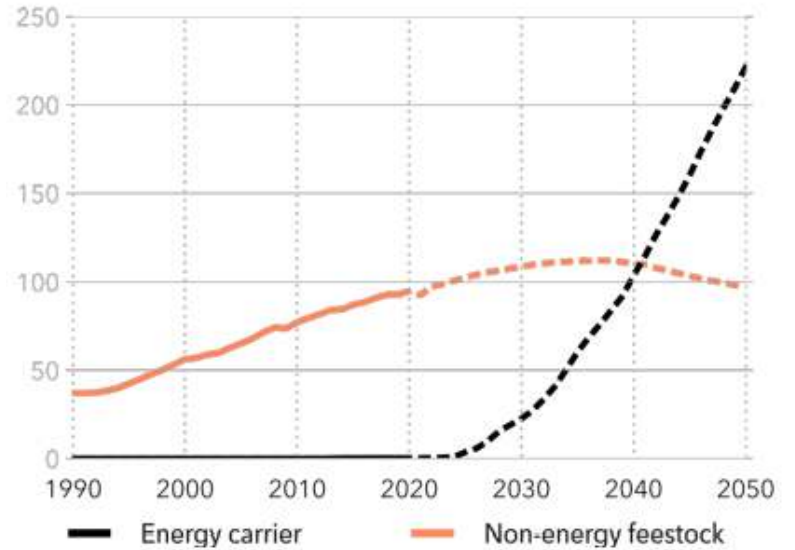
## Weaknesses from the past.....



## ...Opportunities for the future

### Global demand for hydrogen and its derivatives by purpose

Units: Mth<sub>2</sub>/yr



## Understanding the risks of Hydrogen Hydrogen reputation

False starts prejudiced our confidence



...and still influence people opinion



Norway, 2019  
Hydrogen fuelling station explosion



South Korea, 2019  
Hydrogen tank explosion



Illinois, USA, 2019  
Hydrogen blast at AB Speciality Silicones factory

[YouTube - Incompatible Chemicals: Explosion at AB Specialty Silicones](#)

## Understanding the risks of Hydrogen A safety moment



Hydrogen has a much higher burning velocity than hydrocarbons

- More devastating damage in an explosion scenario with concentration above 15%
- Much lighter
  - Easier dispersion but wider ignition
  - High energy per unit of mass
  - Low energy density per unit of volume
  - Boiling point is -252 °C

Property	Hydrogen
Density (gaseous)	0.089 kg/m <sup>3</sup> (0°C, 1 bar)
Density (liquid)	70.79 kg/m <sup>3</sup> (-253°C, 1 bar)
Boiling point	-252.76°C (1 bar)
Energy per unit of mass (LHV)	120.1 MJ/kg
Energy density (ambient cond. LHV)	0.01 MJ/L
Specific energy (liquefied, LHV)	8.5 MJ/L
Flame velocity	346 cm/s
Ignition range	4-77% in air by volume
Autoignition temperature	585°C
Ignition energy	0.02 MJ

*Notes*

cm/s = centimetre per second, kg/m<sup>3</sup> = kilograms per cubic metre,  
LHV = Lower heating value, MJ = Megajoule, MJ/kg = Megajoule per kilogram,  
MJ/L = Megajoule per litre

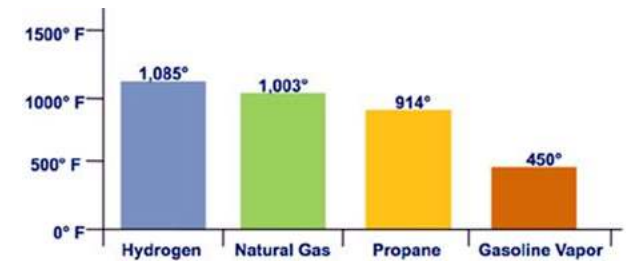
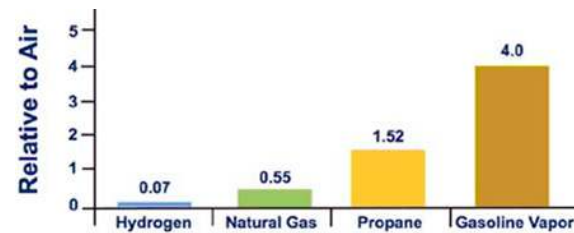
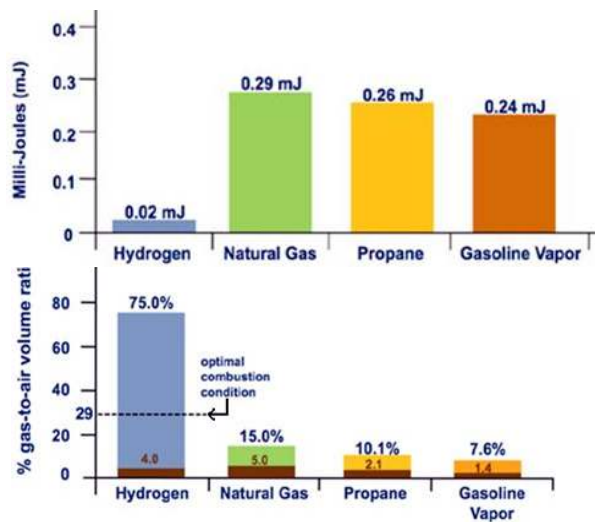
→ Hydrogen is unforgiving



## Understanding the risks of Hydrogen Hydrogen behavior

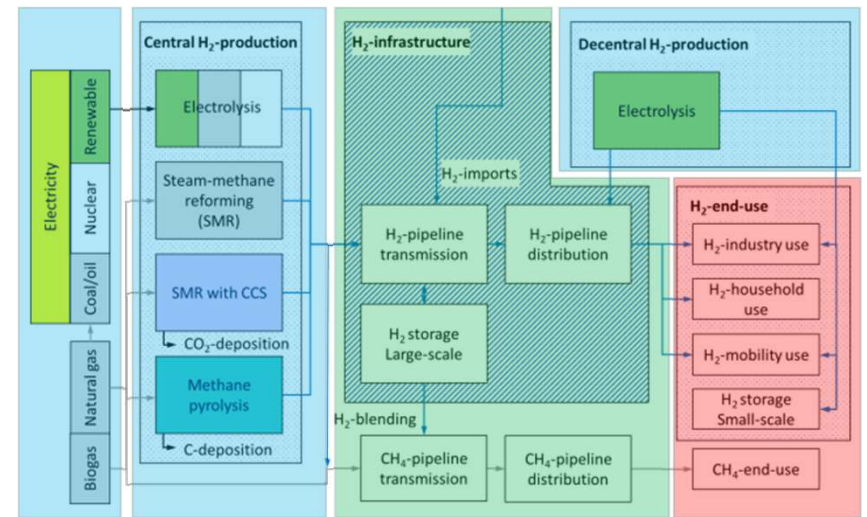
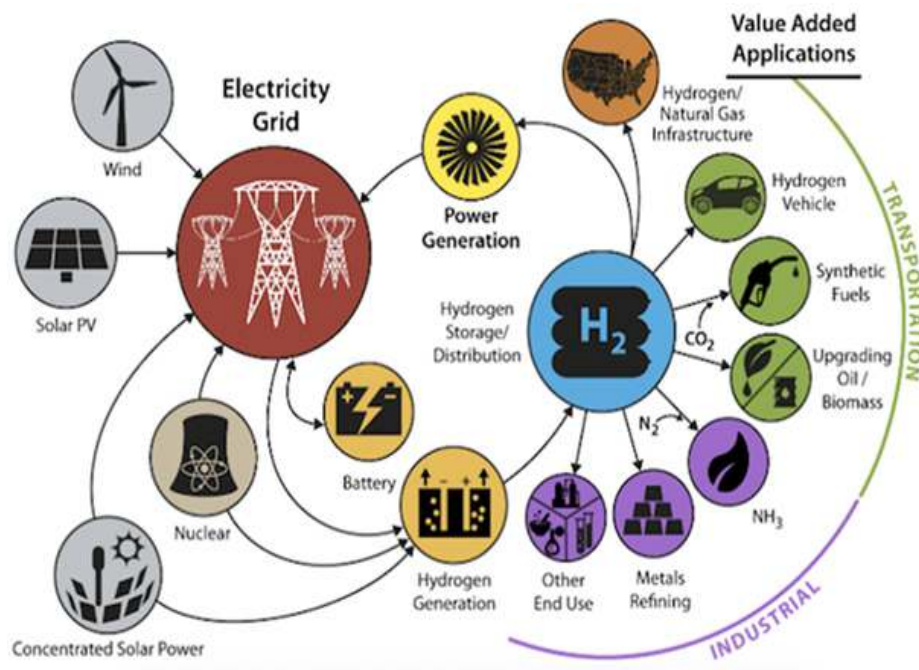
### Comparison with commonly used fuels

- Hydrogen is about 57 times lighter than gasoline vapor → pros and cons depending on the openness of the environment  
→ there are no known odorants light enough to travel with
- Hydrogen has the highest autoignition temperature but the wider flammability range (4 to 75% in air)
- In a situation of optimal combustion conditions, the energy required to initiate hydrogen combustion is much lower than that required for other common fuels



# Understanding the risks of Hydrogen

## Risk exposures and engineering considerations from well-to-end-use



- Source of energy and production technologies
- Distribution and storage
- End users

## Understanding the risks of Hydrogen Donald Rumsfeld Quotes

### Known knowns

Risk exposures are well acquainted and can be assessed for supporting insurers' risk appetite.

### Known unknowns

Awareness of unknown areas of risk pushes for greater attention/deeper engineering investigations and requires a cautious approach from an insurance perspective.

### Unknown unknowns

Risk exposures knowledge is very limited, and this rises barriers to insurers' risk appetite. Adequate expertise and understanding are mandatory to make a fully-fledged risk insurable.

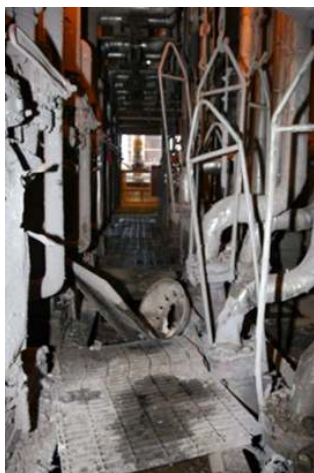


# Understanding the risks of Hydrogen

## «Grey», «Black» or «Blue» Hydrogen: our industry expertise

### «Well-known» hazard

- Steam reforming** • Creep damage or stress corrosion cracking mechanism leading to hydrogen leaks from tubes with potentials for reformer box explosion, refractory damage or hydrogen jet fire from piping rupture
- Coal gasification** • Similar to steam reforming exposures including refractory damage or hydrogen jet fire from piping rupture, but also high-pressure rupture of the gasifier because of the higher operating pressure

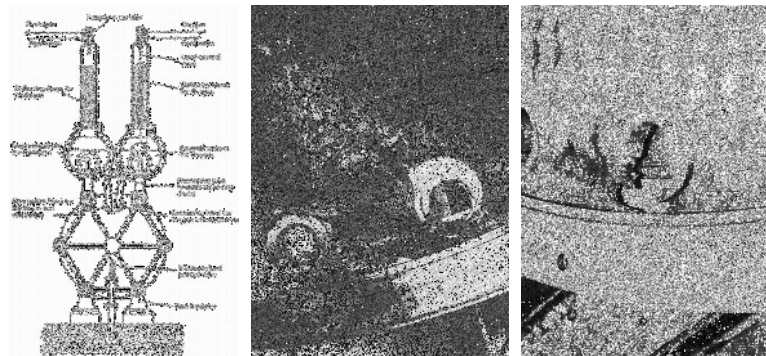


## Understanding the risks of Hydrogen «Green» Hydrogen: an “old” process under continuous development

«Aggravated» exposures • Unknowns include upscaling of the capacities and “new” technologies. Cost reduction means:



- Transition from atmospheric to pressurized Alkaline technology
- Reduction of “rare” and “precious” iridium/platinum in favor of new prototypical materials
- Steam employment for Solid Oxide Electrolysis (SOE)
- Membranes degradation and consequent hydrogen/oxygen contamination remain the driving hazard -> mechanical failures leading to room explosions
- De-oxo units required for further processing are “critical” due to catalyst dusting & runaway reactions

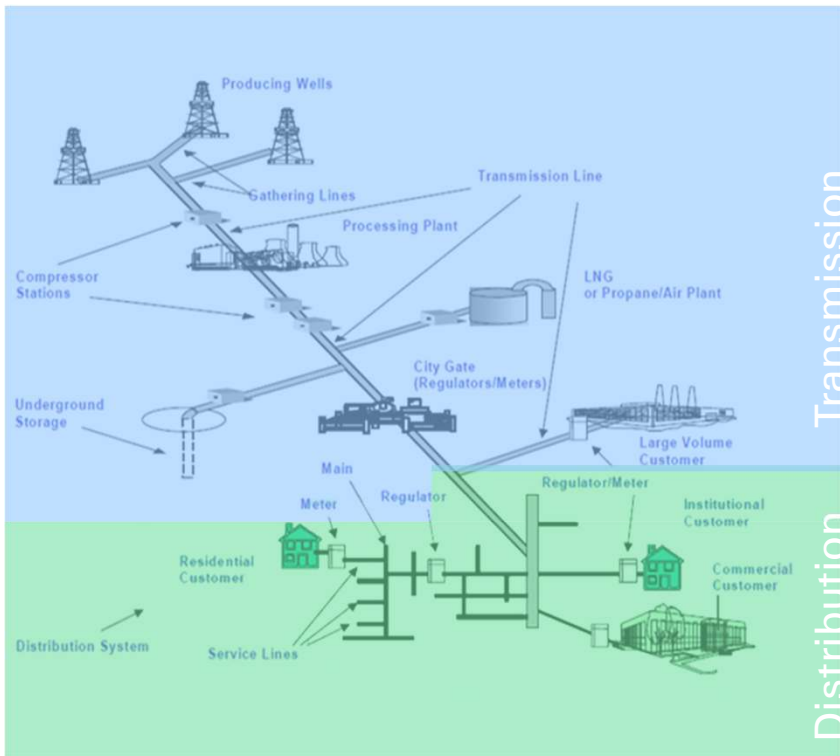


On 5 April 1975, an explosion occurred at the factory of Laporte Industries which resulted in extensive damage to an electrolytor plant and the subsequent death of the plant operator.

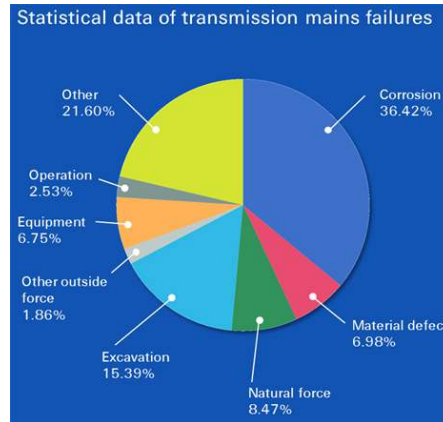


Explosion of an outdoor hydrogen tank at a water electrolyzer plant coupled with a solar photovoltaic power system (May 23th, 2019)  
→ 2 fatalities and 6 injuries (USD 30-40 million damage)

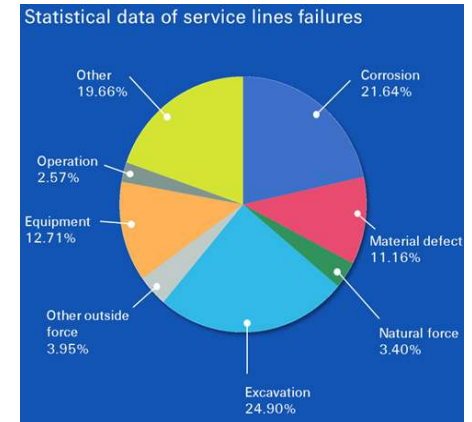
## Understanding the risks of Hydrogen Infrastructure is the «heart»



- Almost consisting of steel pipelines
- Larger size with DN from 4-6 in up to 48 in
- Normal operating pressures in the range of 40-80 barg up to (in some cases) 140 barg
- Loss scenarios dominated by the rupture of the pipeline

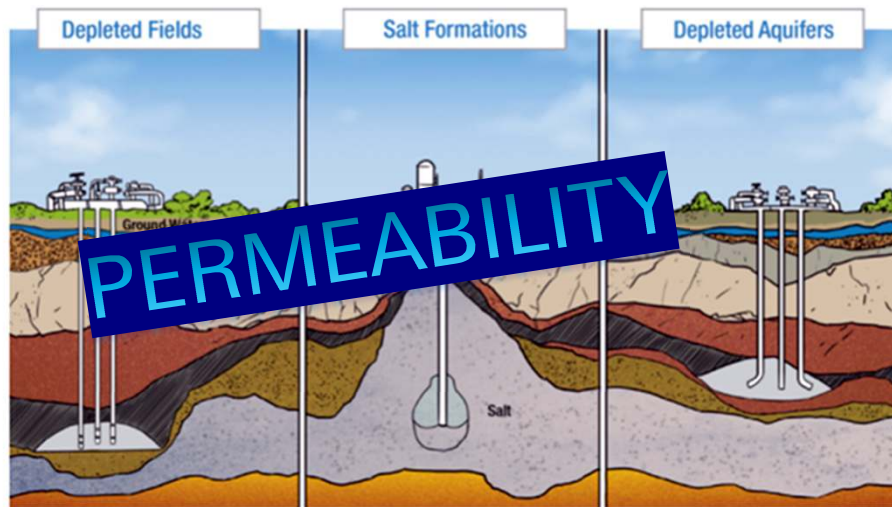


- Low alloy steel and polyethylene are the dominant materials
- Distribution and service lines are typically 1.5 – 4 in
- Normal operating pressures in the range of 1 - 5 barg up in some cases to 10 barg
- Distribution pipeline incidents typically result in leaks



# Understanding the risks of Hydrogen

## Hydrogen storage: risks against efficiency, security and resilience



### Depleted reservoirs

Their imperviousness over geological time periods has already been proven and have already been very well researched (production activities). The suitability criteria include depth and adequate permeability and porosity of the reservoir rock.

- Seasonality
- Capacity

### Salt caverns

Salt caverns are artificial cavities in underground salt formations, which are created by the controlled dissolution of rock salt. They are smaller and considered as one of the best ways to ensure hydrogen purity and tightness.

- Deliverability
- Cost per unit stored

### Aquifers

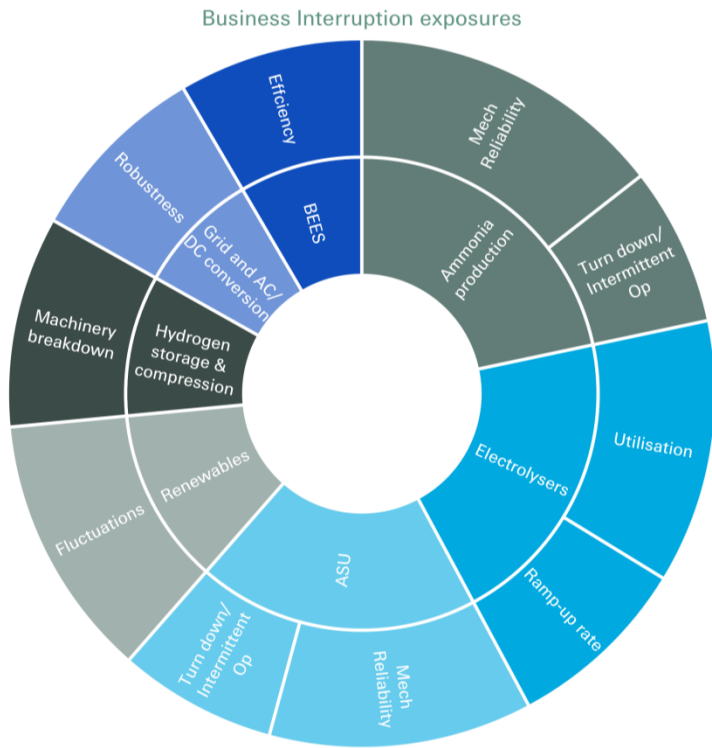
Similar to depleted oil reservoirs in which originally there was a water-fille. Requirements: top sealed by a layer of rock, dome-shaped structure, high permeability with adequate pore space.

- Time to market
- Discharge duration

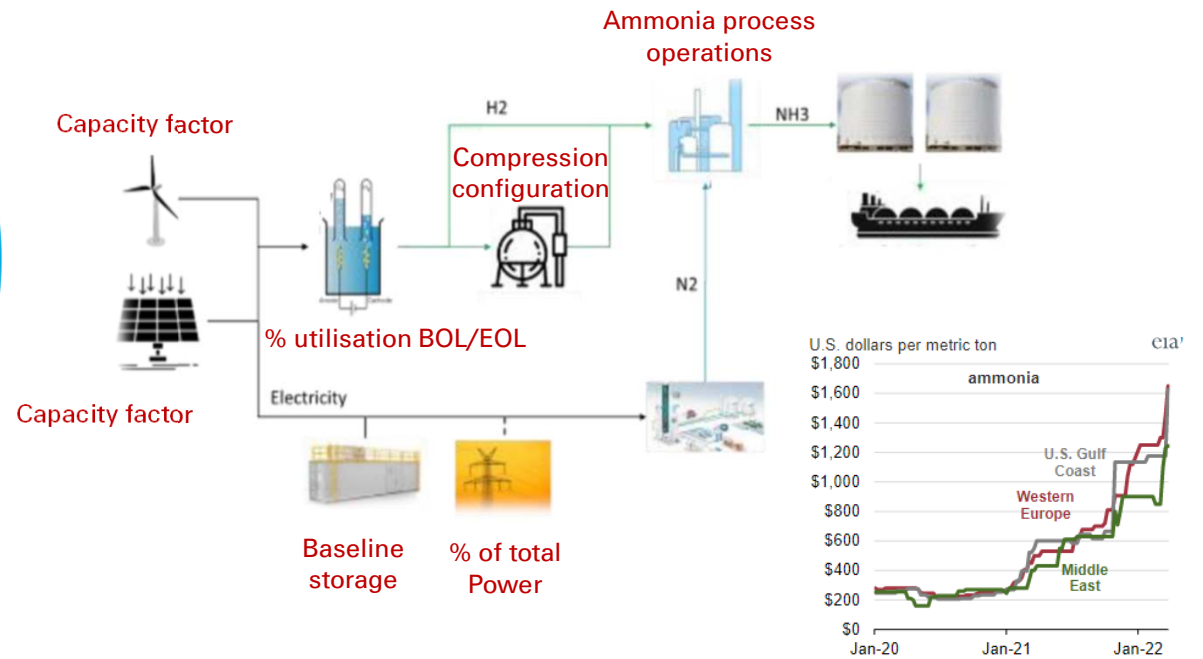
- Energy intensity
- Transportation
- Buoyancy
- Cooling down
- Boil off
- Ignition mechanisms



# Understanding the risks of Hydrogen Business continuity & Mode of operations



## Design Constraints



## Understanding the risks of Hydrogen Knowns and Unknowns

### Known knowns

- Hydrogen production processes from fossils fuels (reforming and gasification)
- Carbon capture & storage
- Traditional electrolyzing technology
- HP (up to 300 bar) storage

### Known unknowns

- Rotating equipment conversions and repurposing
- Upscaling of capacity – R&D in new electrolyzing technology
- Infrastructure retrofitting and high-high pressure storage (700 bar)

### Unknown unknowns

- Cryogenic hydrogen



“Leading in a new economy means patterning our future on the understanding and management of unknown risks. Acting knowingly is better than reacting boldly.”



Massimo Giachino,  
Manager Risk Engineering OPC

# Thank you!

## Contact us



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